

PREPARE GELATIN IN VARIETY OF DISHES

Attractive Desserts Are Relished by Young and Old.

Succulent Fruits and Fruit Ices Are Particularly Acceptable After Comparatively Heavy Meal—Recipe for Pudding.

(Prepared by the United States Department of Agriculture.)

Gelatin can be used in the preparation of a great variety of attractive desserts which are liked by young as well as by old and are for this reason convenient for occasional use in families where adults and children must both be provided for. Gelatin is used in such small amounts that it cannot be relied upon to furnish much nourishment. It takes up a large amount of water, however, and forms bulky or light desserts. Such desserts, which include also succulent fruits and fruit ices, are particularly acceptable after a comparatively heavy meal. Gelatin is a good means of introducing fruit juices, and to some extent also vegetable juices like those of tomatoes and cucumbers, into the diet. These juices contain vitamins, some of which are probably destroyed by heat so that it is well to know a variety of ways of serving them raw. One of these ways is in the form of gelatin desserts. The following recipe is recommended by the United States Department of Agriculture.

Snow Pudding.
1 tablespoonful granulated gelatin, 1 cup cold water, 1 cup boiling water, 1 cup sugar, 1 cup lemon juice, 1 cup cold water, Whites of three eggs.
Soak gelatin in the cold water and then dissolve it in hot water. After adding the sugar and lemon juice, strain and set aside to cool, stirring occasionally. When it is quite thick, beat with an egg beater. Add the well-beaten egg whites and continue to beat until it is stiff enough to hold its shape. Mold or pile by spoonfuls on a glass dish. Serve with a soft custard made with the egg yolks. Variety can be obtained in this dish by substituting fruit juices for all or part of the water. Fruit juices like those of grapes and berries that are not injured in flavor by heat can be used in place of the water for dissolving the gelatin, though this may destroy some of the vitamins.

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PLAN FOR CLEANING SILVER

Common Table Salt and Boiling Water Are Useful—Rub With Good Abrasive Polish.

A satisfactory method for cleaning household silver, according to the United States Department of Agriculture, is as follows: An enamel or agate ware dish should be partly filled with a cleaning solution of one teaspoonful of common table salt to each quart of water and placed directly on the stove to boil. A sheet of aluminum or clean zinc should then be dropped into the dish and the tarnished silver placed in contact with this metal. It is best that the silver be entirely covered with the cleaning solution and that the solution remain at the boiling temperature. As soon as the tarnish has been removed the silver should be removed, rinsed in clean water, and wiped with a soft cloth.

Aluminum corrodes quickly in the cleaning solution, so that aluminum dishes of any value for culinary purposes should never be used. Aluminum ware, that would otherwise be thrown away, or any inexpensive piece of the metal, will serve very satisfactorily for cleaning silver. Zinc may be used in place of aluminum, but it becomes corroded and inactive in a much shorter time. The electrolytic method may be used as frequently as is necessary to remove the tarnish, but it is necessary to rub the silver with some good abrasive polish only as often as may be desirable to restore the burnished appearance.

Value of Slice of Bread.
An ordinary slice of bread (a 1/4-inch slice cut from an ordinary loaf) is equal in food value to about half a cupful of boiled or steamed cereal and to about a cupful of puffed or flaked cereal. The mother who must feed her child very economically should calculate the cost of each and decide which is cheapest, says the United States Department of Agriculture.

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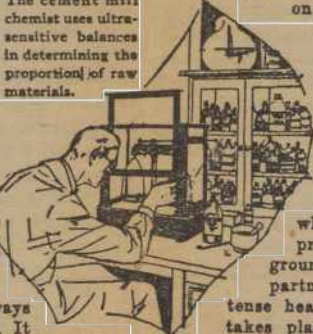


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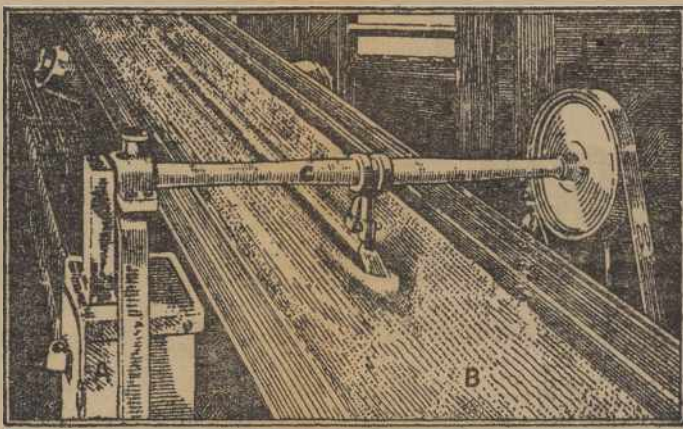
Every eight seconds, mechanical devices dip into the materials that are traveling through a cement mill and put a sample under lock and key for the chemist to test. Chemists are "the czars" of a cement plant and chemistry is their law.

Few factory processes require more care than the making of cement—thousands of test samples being taken in a day's run at a cement plant because cement must be tested in a variety of ways for numerous qualities. It is under the chemist's instructions that workmen proportion the raw materials. These tests in which the proportions of the elements of cement are determined call for glass-enclosed balances or scales that must be accurate to the



ten-thousandth of a gram. There are 454 grams to a pound. Such balances will quickly detect the difference in weight of a piece of paper, before and after you have written your name on it.

The "rule" of the chemist is seen and felt at every point in the process of cement manufacture—in the quarry or pit where raw materials are obtained—in the departments where raw materials are proportioned, mixed and ground—in the burning department where intense heat the chemical change takes place—in the department where the raw material which has now become clinker is ground to the fine powder called portland cement. This country put over thirty million tons of cement making material through these processes last year.



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